

Methyl Ether of De-A,B-8 α -ethynylcholestan-8 β -ol (17). Alcohol 16 (6.9 mmol) and methyl iodide (0.87 mL, 14 mmol) were added to a solution of KOH (1.57 g, 28 mmol) in 14 mL of Me₂SO. The reaction mixture was stirred for 15 min at room temperature, poured in 200 mL of water, and extracted with pentane (2 \times 50 mL). The organic phase was dried over MgSO₄ and evaporated. The crude product was purified by chromatography (hexane/AcOEt, 99/1): yield, 70% from ketone 15; mp 51–52 °C; [α]_D²¹ +4.0° (CHCl₃, c 2.0); IR (CHCl₃) 3310, 2870 cm⁻¹; ¹H NMR (200 MHz, CDCl₃) δ 0.9 (m, 12 H, CH₃), 1–2 (m, 20 H, CH₂ + CH), 2.38 (s, 1 H, CH), 3.31 (s, 3 H, CH₃O); ¹³C NMR (50 MHz, CDCl₃) δ 18.4, 20.9, 23.8, 26.7, 34.4, 35.9, 39.5, 40.3, 42.6, 71.8, 74.7, 76.4, 77.0, and 77.6 (8 CH₂ + 3 C) 13.3, 18.5, 22.5, 22.8, 28.0, 35.4, 51.3, 56.8, and 57.1 (5 CH₃ + 5 CH).

Coupling Reaction of Alkyne 17 with Ketones 6, 11, 13, and 14. General Procedure. *n*-BuLi (1.25 mL, 1.32 M in hexane) was added to 500 mg (1.6 mmol) of product 17 in 5 mL of anhydrous THF. After 5 min at room temperature, 1.6 mmol of ketone in 5 mL of THF was added. After 1 h at room temperature, the reaction mixture was hydrolyzed with 10 mL of saturated NH₄Cl and extracted with ether (2 \times 20 mL). The organic phase was washed with saturated NaCl, dried over MgSO₄, and evaporated. The crude product (80% yield) was used in the next step without purification.

Hydroxy Ethers 18 and 19. The TBDMS ether of 18 or 19 (0.61 mmol) in acetonitrile (10 mL) was treated by 1 mL of 40% FH for 30 min. The solution was then washed with saturated NaHCO₃ and extracted with ether (3 \times 10 mL). The organic layer was washed with saturated NaCl, dried over MgSO₄, and evaporated: quantitative yield; ¹H NMR (60 MHz, CCl₄) δ 0.9–2.1 (m), 3.2 (s, 3 H, OCH₃), 3.6 (m, 1 H, CHOH).

Hydroxy Ethers 20 and 21. ZnBr₂ (1.1 g, 5 mmol) was added to a solution of 1.0 mmol of the MEM ether of 20 or 21 in 20 mL of dichloromethane. The dark reaction mixture was stirred for 8 h and then hydrolyzed with saturated NaHCO₃ (100 mL). The solution became yellow, and a white precipitate formed. The aqueous phase was extracted with CH₂Cl₂ (2 \times 20 mL) and the organic layer dried over MgSO₄ and evaporated: quantitative yield; ¹H NMR (60 MHz, CCl₄) δ 1–2.2 (m), 3.2 (s, 3 H, OCH₃), 3.6 (m, 1 H, CHOH).

Reduction of Propargylic Alcohols 18–21. General Procedure. A titrated LiAlH₄ (2.4 mL, 1.8 mmol, 3 equiv) solution (0.77 M) in ether was added at 0 °C to a solution of 0.61 mmol of propargylic alcohol in 20 mL of ether. After 1 h at 0 °C, the reaction mixture was hydrolyzed with 2 mL of EtOAc and 20 mL of 5% HCl and extracted with ether (3 \times 20 mL). The organic phase was washed with saturated NaCl (100 mL), dried over MgSO₄, and evaporated: quantitative yield; ¹H NMR (200 MHz,

CDCl₃) δ 0.9–2.1 (m), 3.3 (s, 3 H, OCH₃), 3.6 (m, 1 H, CHOH), 4.21 (2 AB systems, 2 H, J_{AB} = 12 Hz, $\Delta\nu$ = 28.6 Hz, vinylic H).

Low-Valent Titanium Reductive Elimination. General Procedure. A titrated solution of LiAlH₄ (8.3 mL, 6.4 mmol) in ether (0.77 M) was added to 1.97 g (12.8 mmol) of TiCl₃ in 10 mL of THF under argon. A black suspension formed. The reaction mixture was stirred at room temperature for 30 min. Then 1.6 mmol of allylic hydroxy ether in 5 mL of THF was added and the reaction mixture refluxed for 1 h, hydrolyzed with 50 mL of 2 N HCl, and extracted with chloroform (3 \times 20 mL). The organic layer was washed with water several times to remove the brown color, dried over MgSO₄, and evaporated.

The diastereoisomer ratio was determined by NMR on the crude product from the signals of the proton α to the hydroxyl group.

The products (DHT₃ and DHTV₃) were then separated by preparative TLC (hexane/ether, 90/10). Yields of purified products after TLC separation are between 60% and 75%.

DHV₃ (5Z,7E,3S,10R)-26: ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 3.50 (m, 1 H, CHOH), 5.9 (m, 2 H, vinylic H).

DHT₃ (5E,7E,3S,10R)-27: [α]_D²¹ +0.5, [α]₃₆₅²¹ +2.4° (c 1.2, acetone); ¹H NMR (200 MHz, CDCl₃) δ -2.5 (m), 3.89 (m, 1 H; CHOH), 6.0 (m, 2 H, vinylic H).

DHV₃ (5Z,7E,3R,10R)-30: ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 3.85 (m, 1 H, CHOH), 5.8 (m, 2 H, vinylic H).

DHT₃ (5Z,7E,3R,10R)-31: [α]_D²¹ +6.3°, [α]₃₆₅²¹ +19.4° (c 2.2, acetone); mp 92 °C; ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 3.65 (m, 1 H, CHOH), 6.61 (AB, J_{AB} = 11 Hz, $\Delta\nu$ = 33 Hz, vinylic H).

DHV₃ (5Z,7E,3R,10S)-28: ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 3.66 (m, 1 H, CHOH), 6.16 (AB, 2 H, J_{AB} = 10 Hz, $\Delta\nu$ = 7.5 Hz, vinylic H).

DHT₃ (5E,7E,3R,10S)-29: [α]_D²¹ +1.2°, [α]₃₆₅²¹ +4.8° (c 1.1, acetone); ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 3.73 (m, 1 H, CHOH), 6.24 (AB, 2 H, J_{AB} = 11.5 Hz, $\Delta\nu$ = 7 Hz).

DHV₃ (5Z,7E,3S,10S)-32: [α]_D²¹ +36.9; [α]₃₆₅²¹ +152° (c 0.4, acetone); ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 4.00 (m, 1 H, CHOH), 5.9 (m, 2 H, vinylic H).

DHT₃ (5E,7E,3S,10S)-33: [α]_D²¹ +5.2°, [α]₃₆₅²¹ +14.8° (c 1.1, acetone); ¹H NMR (200 MHz, CDCl₃) δ 1–2.5 (m), 3.65 (td, 1 H, CHOH), 6.03 (AB, 2 H, J_{AB} = 10 Hz, $\Delta\nu$ = 30 Hz, vinylic H).

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Synthetic Approaches to Nogalamycin-Related Anthracyclines. An Approach to a Western Synthone¹

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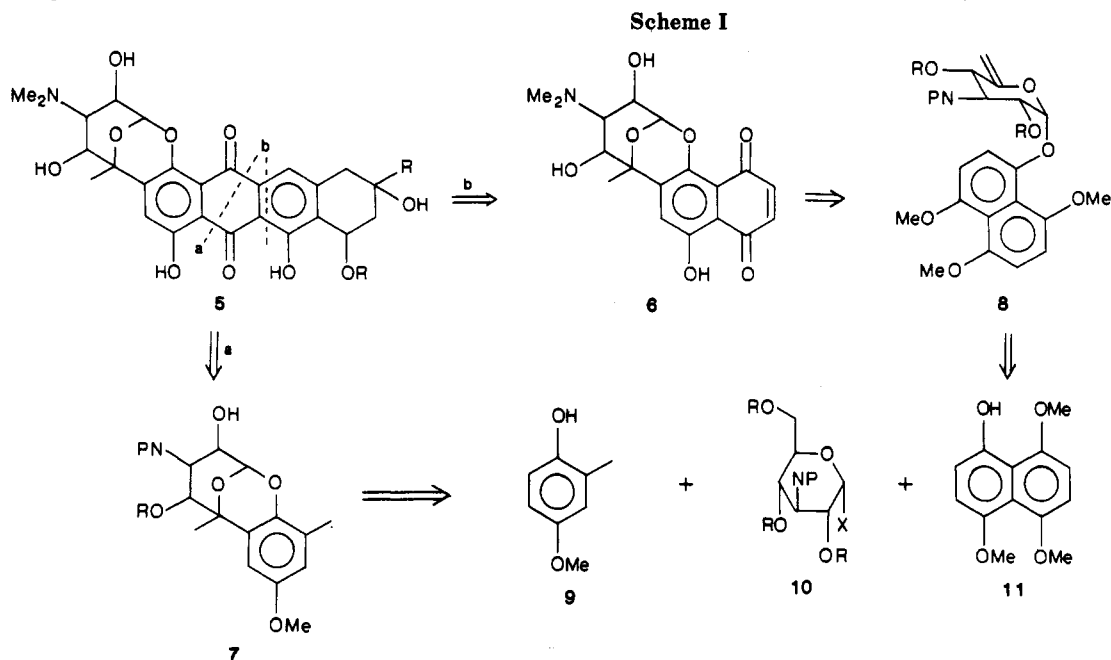
A synthetic approach to the "western" portion of nogalamycin is described. The synthesis of the 3-aminoglucose synthon 21 and its utilization in the stereocontrolled glycosidation of the D-ring synthon 9 is reported. The resulting glycoside 24 was converted to the 5',6'-olefin 36. The formation of the 2,5'-C–C bond was attempted via an intramolecular cation induced olefinic cyclization approach. While model intermolecular studies were successful, the intramolecular reaction failed presumably due to insufficient nucleophilicity of the aromatic moiety.

The clinical utility of the anthracycline antibiotics daunorubicin (1) and doxorubicin (2)² in the treatment of cancer has prompted the development of so-called second

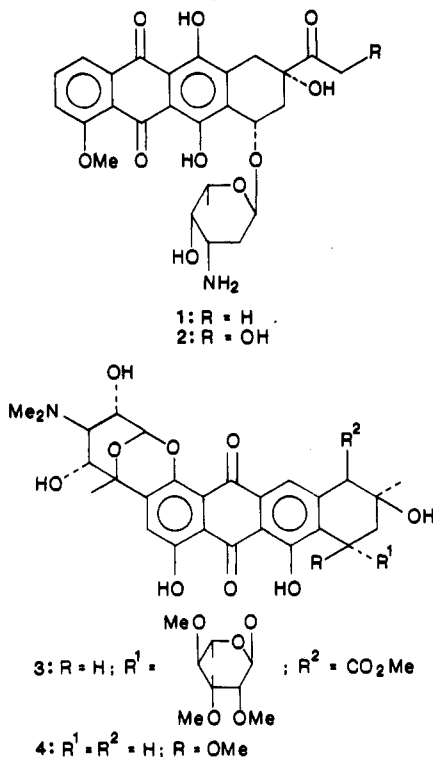
generation anthracyclines which include both natural products and their semisynthetic analogues. Therapeutic

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(1) A preliminary account of this work was presented by T. H. Smith and H. Y. Wu at the 185th National Meeting of the American Chemical Society, Seattle, WA, March, 1983, Abstract ORGN 14.



advantages over 1 and 2 in experimental tumors have been claimed for several of these agents. Nogalamycin (3) and its derivatives are examples of such second generation drugs isolated from *Streptomyces nogalater* var. *nogalater* sp.n. The structure,³ except for A-ring stereochemistry



and configuration of the aminoglucose residue, and biosynthesis⁴ of 3 were determined by Wiley et al. More recent studies indicate the complete structure to be as shown in which the aminoglucose moiety has the L-configuration and the 9-hydroxyl has a β -orientation unlike that found in other anthracyclines.⁵ From a program of

structure modification 4 has emerged as the most interesting analogue of 3 and appears to have potential as a clinically useful antitumor agent. Nogalamycin and its analogs are structurally different from the classical anthracyclines in that the amino sugar residue is joined to the aromatic D-ring via both glycoside and C-C bonds forming a benzoxocin ring system. Additional examples of anthracyclines containing this structural feature have recently been reported.⁷ In this paper we report our efforts to prepare a synthon incorporating the "western" portion of nogalamycin including the unusually fused amino sugar residue suitable for eventual elaboration to nogalamycin related anthracyclines.

Due to the lack of appropriate synthetic methodology for the construction of the optically active aminoglucose-2,6-epoxy-2H-1-benzoxocin moiety at the outset of this study, we focused our initial efforts on the "western" portion of nogalamycin.⁸ Our general approach is outlined retrosynthetically in Scheme I. Our plan involves (1) syntheses of a suitably functionalized aminoglucose and a CD- or D-ring synthon, (2) stereoselective glycosidation of the D-ring-containing fragment with the amino sugar reagent, and (3) adjustment of the functionality of the resulting glycoside to permit formation of the bond between C-2 and C-5'. Further elaboration of the western synthon then becomes a problem of 11-deoxyanthracycline synthesis which has been successfully addressed by several groups in recent years.⁹

(5) (a) Wiley, P. F. *J. Nat. Prod.* 1979, 42, 569. (b) Wiley, P. F.; Elrod, D. W.; Houser, D. J.; Johnson, J. L.; Pshchigoda, L. M.; Kreuger, W. C.; Moscovitz, A. *J. Org. Chem.* 1979, 44, 4030. (c) Eckle, E.; Stężowski, J. J.; Wiley, P. F. *Tetrahedron Lett.* 1980, 21, 507. (d) Wiley, P. F.; Elrod, D. W.; Richard, F. A. *J. Med. Chem.* 1982, 25, 560. (e) Arora, S. K. *J. Am. Chem. Soc.* 1983, 105, 1328.

(6) (a) Wiley, P. F.; Johnson, J. L.; Houser, D. J. *J. Antibiot.* 1977, 30, 628. (b) McGovern, J. P.; Nelson, K. G.; Lassus, M.; Craddock, J. C.; Plowman, J.; Christopher, J. P. *Invest. New Drugs* 1984, 2, 359.

(7) (a) Ishi, K.; Kondo, S.; Nishimura, Y.; Hamada, M.; Takeuchi, T.; Umezawa, H. *J. Antibiot.* 1983, 36, 451. (b) Kawai, H.; Hayakawa, Y.; Nakagawa, M.; Imamura, K.; Tanabe, K.; Shimazu, A.; Seto, H.; Otake, N. *J. Antibiot.* 1983, 36, 1569. (c) Kawai, H.; Hayakawa, Y.; Nakagawa, M.; Furihata, K.; Seto, H.; Otake, N. *Tetrahedron Lett.* 1984, 25, 1941.

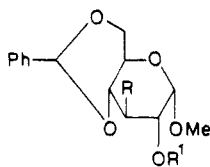
(8) Several partial solutions to this problem have recently been reported. (a) Bates, M. A.; Sammes, P. G. *J. Chem. Soc., Chem. Commun.* 1983, 896. (b) Hauser, F. M.; Ellenberger, W. P.; Adams, T. C., Jr. *J. Org. Chem.* 1984, 49, 1169. (c) Hauser, F. M.; Adams, T. C., Jr. *J. Org. Chem.* 1984, 49, 2296. (d) Kawasaki, M.; Matsuda, F.; Terashima, S. *Tetrahedron Lett.* 1985, 26, 2693.

(2) (a) Henry, D. W. In *Cancer Chemotherapy*; Sartorelli, A. C., Ed.; ACS Symposium Series 30; American Chemical Society: Washington, DC, 1976; pp 15-57. (b) Arcamone, F. *Med. Res. Rev.* 1984, 4, 153.

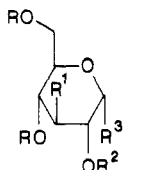
(3) Wiley, P. F.; Kelly, R. B.; Caron, E. L.; Wiley, V. H.; Johnson, J. H.; MacKellar, F. A.; Miszak, S. A. *J. Am. Chem. Soc.* 1977, 99, 542.

(4) Wiley, P. F.; Elrod, D. W.; Marshall, V. P. *J. Org. Chem.* 1978, 43, 3457.

The partially protected 3-aminoglucose derivative **12**,¹⁰ readily available from D-glucose, was a convenient starting material for the preparation of the amino sugar synthon.¹¹ The amino group was protected by conversion (*N*-carboethoxyphthalimide, 91%) to the phthalimide **13**. The 2-hydroxyl was protected via methylation (MeI, Ag₂O, CH₂Cl₂, 76%), which afforded **14**. Protection of the 4- and 6-hydroxyls to the acidic conditions required to functionalize the anomeric carbon for glycosidation was achieved by mild acid hydrolysis (50% HOAc) of the benzylidene moiety followed by *p*-nitrobenzoylation (*p*-O₂NC₆H₄COCl, DMAP, 77%) to provide **17**. Functionalization of the anomeric carbon was accomplished via acetolysis of **17** (Ac₂O, HOAc, H₂SO₄, 92%) to provide **18**. Reaction of **18** with anhydrous HCl or HBr afforded the corresponding halo sugars **19** and **20**, respectively. Acetolysis of **14** afforded **21** (74%) thus functionalizing the 1-position and affording acid-stable protection to the 4- and 6-hydroxyls in a single operation.



- 12**: R = NH₂; R¹ = H
13: R = NPht; R¹ = H
14: R = NPht; R¹ = Me
15: R = NHTFA; R¹ = H
16: R = NHTFA; R¹ = Bzl



- 17**: R = PNB; R¹ = NPht; R² = Me; R³ = OMe
18: R = PNB; R¹ = NPht; R² = Me; R³ = OAc
19: R = PNB; R¹ = NPht; R² = Me; R³ = Cl
20: R = PNB; R¹ = NPht; R² = Me; R³ = Br
21: R = Ac; R¹ = NPht; R² = Me; R³ = OAc
22: R = PNB; R¹ = NHTFA; R² = Bzl; R³ = OMe

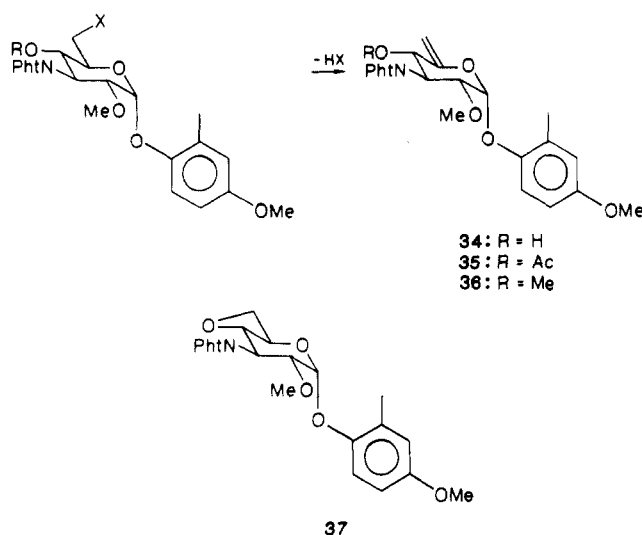
The choice of protecting groups in this sequence proved to be crucial. The protected aminoglucose derivative **22**, prepared from **13** by standard methods via **15** and **16**, was unsuitable as a synthon for incorporating the amino sugar

(9) (a) Bauman, J. G.; Barber, R. B.; Gless, R. D.; Rapoport, H. *Tetrahedron Lett.* **1980**, *21*, 4777. (b) Tatsuta, K.; Takeuchi, T. *J. Antibiot.* **1980**, *33*, 1581. (c) Confalone, P. N.; Pizzalato, G. *J. Am. Chem. Soc.* **1981**, *103*, 4251. (d) Boeckman, R. K., Jr.; Seem, F.-W. *J. Am. Chem. Soc.* **1982**, *104*, 4604. (e) Mitscher, L. A.; Wu, T.-S.; Khanna, I. *Tetrahedron Lett.* **1983**, *24*, 4809. (f) Vedejs, E.; Miller, W. H.; Pribish, J. R. *J. Org. Chem.* **1983**, *48*, 3611. (g) Rama Rao, A. V.; Chanda, B. M.; Borate, H. B. *Synth. Commun.* **1984**, *14*, 257. (h) Deshpande, V. H.; Ravichandran, K.; Ramamohan Rao, B. *Synth. Commun.* **1984**, *14*, 477. (i) Uemura, M.; Minami, T.; Hayashi, Y. *J. Chem. Soc., Chem. Commun.* **1984**, 1193. (j) Krohn, K. *Liebigs Ann. Chem.* **1983**, 2151. (k) Kelly, T. R., Ed. *Tetrahedron (Symposia-in-Print)* **1984**, *40* (17), 4537-4789. (l) Tanaka, H.; Yoshioka, T.; Shimauchi, Y.; Yoshimoto, A.; Ishikura, T.; Naganawa, H.; Takeuchi, T.; Umezawa, H. *Tetrahedron Lett.* **1984**, *25*, 3355 and references cited therein. (m) Bauman, J. G.; Hawley, R. C.; Rapoport, H. *J. Org. Chem.* **1985**, *50*, 1569.

(10) Guthrie, R. D.; Johnson, L. F. *J. Chem. Soc.* **1961**, 4166.

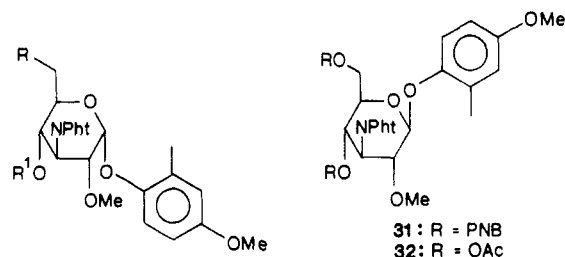
(11) When this study was initiated the configuration of the amino glucose residue had not been determined. We chose to work with the D-glucose series due to its greater accessibility and biosynthesis evidence (ref 4) that the amino sugar of **3** was of the D series. X-ray data has now been reported (ref 4e) indicating that the amino sugar has the L configuration. However, the chemistry developed for the D series should also be applicable to the corresponding compounds having the L-glucose configuration.

Scheme II

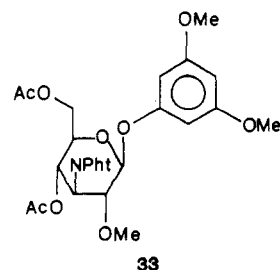


residue into the target. The main obstacle to the use of **22** was the resistance of the methyl glycoside to hydrolysis under conditions to which the 2-*O*-benzyl and 3-trifluoroacetamido groups were stable.

Glycosidation of **9**¹² and **11**¹³ with the halo sugars **19** and **20** could not be achieved under a variety of Koenigs-Knorr conditions. An additional complication was the instability of the CD-ring synthon **11** under these conditions. This led us to concentrate on **9** as a simpler substrate upon which to work out suitable glycosidation conditions. The SnCl₄-catalyzed¹⁴ reactions of **9** with **18** and **21** provided the α -glycosides **23** and **24** in reasonable yield along with the corresponding β -anomers **31** and **32** in a 5:1 ratio. The



- 23**: R = R¹ = OPNB
24: R = R¹ = OAc
25: R = R¹ = OH
26: R = OMs; R¹ = OH
27: R = I; R¹ = OH
28: R = I; R¹ = OAc
29: R = OMs; R¹ = OMe
30: R = I; R¹ = OMe

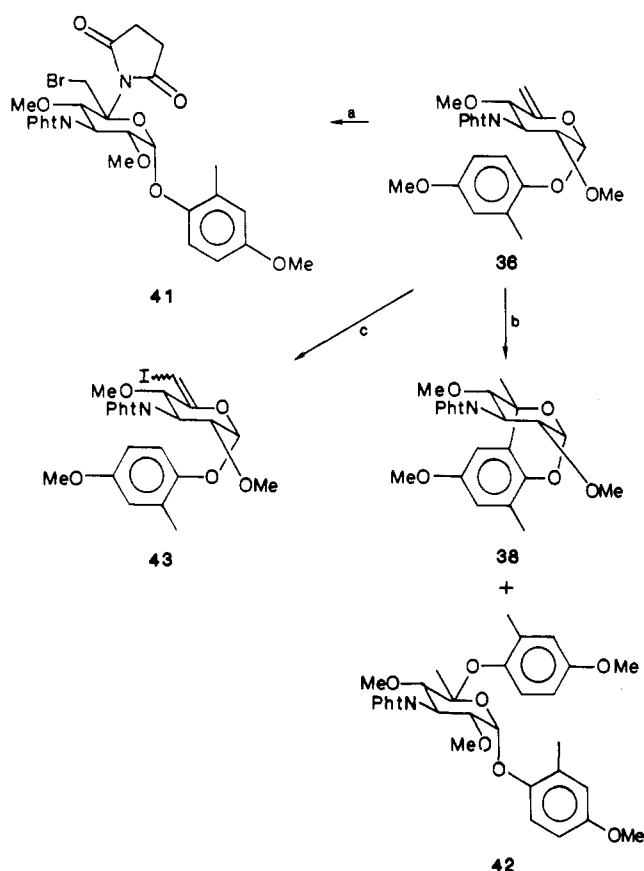


stereocontrol achieved in this process was presumed to be

(12) Green, J.; McHale, D.; Mammalis, P.; Marcinkewicz, S. *J. Chem. Soc.* **1959**, 3374.

(13) Smith, T. H.; Wu, H. Y. *J. Org. Chem.* **1982**, *47*, 1974.

(14) Honma, K.; Nakazima, K.; Uematsu, T.; Hamada, A. *Chem. Pharm. Bull.* **1976**, *24*, 394.

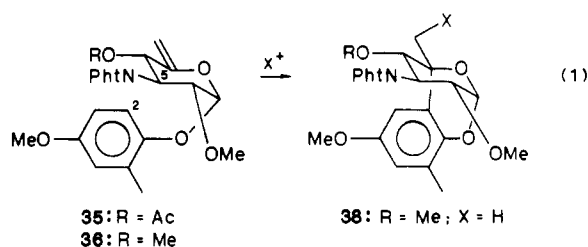
Scheme III^a

^a (a) NBS; (b) TFA; (c) iodonium dicollidine perchlorate.

due to the influence of the phthalimido group. Steric control of a glycosidation reaction by a phthalimido moiety has previously been reported.¹⁵ However an attempt to extend this process to the condensation of 3,5-dimethoxyphenol with 21 afforded only the β -glycoside 33. An explanation for the dramatic difference in the stereochemical outcome of these seemingly similar reactions is not obvious.

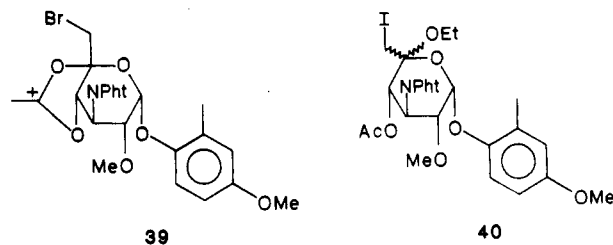
With the α -glycoside in hand, our next objective was to generate the 5',6'-olefin required for the projected cyclization to form the 2,5'-C-C bond (Scheme II). Quantitative removal of the *O*-acyl protecting groups of 23 and 24 was achieved with Dowex resin (OH⁻ form) in EtOH and Na₂CO₃ in MeOH, respectively, to provide 25. Selective functionalization of the 6-position was best achieved via reaction of 25 with 1 equiv of methanesulfonyl chloride in pyridine to afford 26. Reaction of 26 with NaI in acetone afforded 27 in high yield. It ultimately proved to be necessary to protect the 4'-hydroxyl. This could be achieved either by acylation (Ac₂O, DMAP, CH₂Cl₂) of 27 to provide 28 or by methylation (MeI, Ag₂O, CH₂Cl₂) of 26 to provide 29, which is converted to 30 via reaction with NaI in 2-butanone. Attempts to generate the olefin 34 from 26 or 27 were unsuccessful, possibly due to the presence of the free hydroxyl, which in similar systems can form oxetanes such as 37 via intramolecular displacement of the leaving group.¹⁶ Olefins 35 and 36 were readily prepared via reaction of the hydroxyl-protected intermediates 28 and 30, respectively, with DBU in HMPA.¹⁷

Having synthesized 35, our next objective was the formation of the C-C bond between C-5 of the amino sugar residue and C-2 of the D-ring synthon. To achieve ring closure we proposed to develop a positive charge at C-5 of the vinyl ether which would then undergo intramolecular nucleophilic attack by the electron rich aromatic system (eq 1). Our approach was to treat the olefin with



a proton source or equivalent, such as a Lewis acid or positive halogen species, which could generate the positive charge at C-5'.

The mode of protection of the 4'-hydroxyl proved to be critical in this process. Reaction of 35 with *N*-bromosuccinimide (NBS), iodonium dicollidine perchlorate,¹⁸ or a catalytic amount of TFA gave rise to a less polar (TLC) product which could not be isolated but decomposed on workup to an intractable mixture of polar products. NMR monitoring of the reaction between 35 and NBS revealed the rapid disappearance of the 4'-acetate signal. This is consistent with interaction of the 4'-acetate moiety with the positive center at C-5' giving rise to a species such as 39, which could decompose to the observed polar products.



However it was possible to trap the C-5' cation. Reaction of 35 with iodonium dicollidine perchlorate in the presence of an external nucleophile, EtOH, gave rise to 40. Although this result supported the general feasibility of our method, it was clear that a nonparticipating protecting group for the 4'-hydroxyl was required if this approach was to be successful. This led us to investigate 36 as a cyclization substrate (Scheme III). Reaction of 36 with NBS was unsuccessful, generating an unstable, less polar product characterized as 41 by NMR. However, reaction of 36 with a catalytic amount of TFA afforded, after workup, mostly starting material along with low and erratic yields of a material tentatively characterized as the desired product 38 as well as the bisaglycone adduct 42. Reaction of 36 with iodonium dicollidine perchlorate afforded the iodo olefin 43 as the major product.

These results indicate that the reaction of 36 at least under the conditions studied thus far, follows the pathway outlined in eq 2. Positively charged species do react with the olefin to generate a carbocation at C-5'. However, proton loss to afford an olefinic species is apparently favored over nucleophilic attack by the aromatic moiety.

Future research in this area could explore the reaction of 36 with other Lewis acid catalysts and the cyclization of olefinic glycosides such as 44 and 45 in which the nu-

(15) Ogawa, T.; Nakahayashi, S.; Sasajima, K. *Carbohydr. Res.* 1981, 95, 308.

(16) Hough, L.; Otter, B. *J. Chem. Soc., Chem. Commun.* 1966, 173.

(17) Crugnola, A.; Lombardi, P.; Gandolfe, C.; Arcamone, F. *Gazz. Chem. Ital.* 1981, 111, 395.

(18) (a) Lemieux, R. U.; Morgan, A. R. *Can. J. Chem.* 1965, 43, 2199. (b) Pauls, H. W.; Fraser-Reid, B. *J. Am. Chem. Soc.* 1980, 102, 3956.

p-nitrobenzoyl chloride (29.3 g, 158 mmol), triethylamine (16 mL), and 4-(dimethylamino)pyridine (6.9 g, 56.4 mmol) were placed in DMF (300 mL) and stirred at 0 °C for 15 h. The mixture was treated with water (10 mL), stirred for 0.5 h at 23 °C, diluted with water (500 mL), and extracted with CH₂Cl₂ (3 × 500 mL). The extracts were combined, washed with 1 N HCl (2 × 500 mL), saturated NaHCO₃ (500 mL), and saturated NaCl (200 mL), dried, and evaporated. The residue was chromatographed (dry column, silica gel, 30 mm × 5 ft, 98:2 CH₂Cl₂/MeOH) and the product recrystallized from CH₂Cl₂/hexane to afford 23.6 g (77%) of 17: mp 202–203 °C; IR 5.60, 5.75 (C=O), 6.18, 6.51, 6.80, 7.22, 7.40, 7.76, 9.00, 9.45, 11.40, 12.00, 12.74, 13.87 μm; NMR δ 3.37 (s, 3, 2-OMe), 3.58 (s, 3, 1-OMe), 4.2–4.7 (m, 4), 4.94 (t, 1, 3-H), 5.12 (d, 1, *J* = 3 Hz, 1-H), 6.00 (t, 1, 4-H), 7.6–7.9 (m, 4, Pht H's), 7.9–8.4 (m, 8, PNB H's); UV (MeOH) λ_{max} 219 nm (ε 51 000), 234 (23 300), 242 (29 400), 258 (27 100); TLC (1:1 hexane/EtOAc) *R_f* 0.50.

Anal. Calcd for C₃₀H₂₅N₃O₁₃: C, 56.69; H, 3.97; N, 6.61. Found: C, 56.36; H, 4.07; N, 6.53.

Acetyl 3-Deoxy-2-methoxy-4,6-bis-*O*-(*p*-nitrobenzoyl)-3-phthalimido- α -D-glucopyranoside (18). To a solution of 17 (23.0 g, 36.2 mmol) and acetic anhydride (135 mL) in glacial acetic acid (720 mL) at 10 °C was added dropwise concentrated H₂SO₄ (43.2 mL). The solution was kept at 23 °C for 48 h and then poured into ice-water (4 L). The precipitate was collected, dried, and recrystallized from CH₂Cl₂/hexane to afford 22.0 g (92%) of 18: mp 165–166 °C; IR 5.80, 5.90, 6.25, 6.60, 6.90, 7.30, 7.50, 7.95, 8.25, 8.75, 9.10, 9.90, 10.70, 11.00, 11.50, 12.05, 12.80, 13.97 μm; NMR δ 2.28 (s, 3, 1-OAc), 3.34 (s, 3, 2-OMe), 4.3–4.7 (m, 4), 4.88 (t, 1, 3-H), 6.05 (t, 1, 4-H), 6.60 (d, 1, *J* = 3 Hz, 1-H), 7.6–7.9 (m, 4, Pht H's), 7.9–8.4 (m, 8, PNB H's); UV (MeOH) λ_{max} 258 nm (ε 27 230), 242 (25 500), 233 (23 400), 219 (52 100); TLC (2:1 hexane/EtOAc) *R_f* 0.2.

Anal. Calcd for C₃₁H₂₅N₃O₁₄: C, 56.11; H, 3.80; N, 6.33. Found: C, 55.88; H, 4.27; N, 6.07.

Bromo 3-Deoxy-2-methoxy-4,6-bis-*O*-(*p*-nitrobenzoyl)-3-phthalimido- α -D-glucopyranoside (20). Anhydrous HBr was passed into a solution of 18 (1.40 g, 2.1 mmol) and acetyl bromide (2.0 mL) in CH₂Cl₂ (25 mL) at 23 °C for 0.5 h. The solution was allowed to stand at 23 °C for 16 h and evaporated. Two 10-mL portions of benzene and CH₂Cl₂ were added to and evaporated from the residue, which was then dried under vacuum to afford 1.45 g (100%) of 20: IR 5.65, 5.80, 5.85 (C=O), 6.25, 6.60, 6.88, 6.95, 7.25, 7.30, 7.48, 7.62, 7.90, 8.00, 8.40, 8.72, 9.10, 9.68, 9.88, 11.46, 11.65, 12.00, 12.80, 13.95, 14.65 μm; NMR δ 3.37 (s, 3, 2-OMe), 4.40–4.75 (m, 4), 4.96 (t, 1, 3-H), 6.05 (t, 1, 4-H), 6.80 (d, 1, *J* = 3 Hz, 1-H), 7.35 (s, 2, C₆H₆), 7.6–7.9 (m, 4, Pht H's), 7.9–8.4 (m, 8, PNB H's); TLC (2:1 hexane/EtOAc) *R_f* 0.35.

Anal. Calcd for C₂₉H₂₂BrN₃O₁₂·¹/₃ C₆H₆: C, 52.40; H, 3.40; N, 5.91. Found: C, 52.74; H, 3.58; N, 5.91.

Acetyl 4,6-Di-*O*-acetyl-3-deoxy-2-methoxy-3-phthalimido- α -D-glucopyranoside (21). To a solution of 14 (42.0 g, 98.7 mmol) and acetic anhydride (250 mL) in glacial acetic acid (1.3 L) at 10 °C was added dropwise concentrated H₂SO₄ (79 mL). The solution was stirred at 23 °C under N₂ for 72 h. The reaction mixture was poured into ice-water (4 L) and extracted with EtOAc (4 × 1 L). The extracts were combined, washed with water (500 mL) and saturated NaHCO₃ (2 × 1 L), dried, and evaporated. The residue was crystallized from CH₂Cl₂/hexane to afford 40 g (90%) of 21: IR 5.72, 5.85, 6.95, 7.35, 8.00, 8.28, 8.75, 9.08, 9.30, 9.60, 9.90, 10.67, 10.90, 11.25, 12.50, 13.86 μm; NMR δ 1.87 and 2.03 (2 s, 6, 4- and 6-OAc's), 2.22 (s, 3, 1-OAc), 3.30 (s, 3, 2-OMe), 4.18 (m, 4), 4.57 (m, 1), 5.55 (t, 1, 4-H), 6.52 (d, 1, *J* = 3 Hz, 1-H), 7.80 (m, 4, Pht H's); TLC (2:1 hexane/EtOAc) *R_f* 0.25.

Anal. Calcd for C₂₁H₂₃NO₁₀: C, 56.12; H, 5.16; N, 3.12. Found: C, 56.13; H, 5.08; N, 3.03.

4-Methoxy-2-methylphenyl 3-Deoxy-2-methoxy-4,6-bis-*O*-(*p*-nitrobenzoyl)-3-phthalimido- α -D-glucopyranoside (23). SnCl₄ (0.45 mL) was added to a solution of 9 (414 mg, 3.0 mmol) and 18 (1.0 g, 1.50 mmol) in CH₂Cl₂ (30 mL). The mixture was stirred at 23 °C for 5 h, diluted with CH₂Cl₂ (150 mL), washed with water (100 mL), dried, and evaporated. While it is more convenient to carry this material on to the next step it can be purified at this stage by chromatography (preparative HPLC, silica gel, 3:1 hexane/EtOAc) to afford 350 mg (32%) of an 86:14 mixture of 23 and the β -anomer 31: IR 5.64, 5.80, 6.23, 6.60, 6.70, 6.90, 7.30, 7.48, 7.65, 7.90, 8.30, 9.00, 9.10, 9.75, 9.90, 10.65, 11.45, 11.55,

11.72, 12.80, 13.95 μm; NMR δ 2.39 (s, 3, ArMe), 3.37 (s, 3, 2-OMe), 3.75 (s, 3, ArOMe), 4.4–4.6 (m, 3), 4.62 (dd, 1, *J* = 10 Hz and 3 Hz, 2-H), 5.15 (t, 1, 3-H), 5.70 (d, 1, *J* = 3 Hz, 1-H), 6.05 (t, 1, 4-H), 6.55 (dd, 1, Ar H), 6.78 (d, 1, Ar H), 7.14 (d, 1, Ar H), 7.6–7.9 (m, 4, Pht H's), 7.9–8.4 (m, 8 PNB H's); MS, *m/e* (relative intensity) 741 [M⁺] (1), 604 (8), 437 (25), 290 (48), 270 (100), 150 (100); UV (MeOH) λ_{max} 259 nm (ε 27 400), 242 (25 500), 219 (61 300); TLC (2:1 hexane/EtOAc) *R_f* 0.35.

Anal. Calcd for C₃₇H₃₁N₃O₁₄: C, 59.91; H, 4.22; N, 5.66. Found: C, 59.50; H, 4.13; N, 5.66.

4-Methoxy-2-methylphenyl 4,6-Di-*O*-acetyl-3-deoxy-2-methoxy-3-phthalimido- α -D-glucopyranoside (24). SnCl₄ (1.15 mL) was added to a solution of 21 (3.4 g, 7.57 mmol) and 9 (1.26 g, 9.08 mmol) in CH₂Cl₂ (35 mL) at 23 °C and the solution was stirred for 2.5 h. A second portion of 9 (105 mg, 0.76 mmol) was added and stirring was continued for an additional 0.5 h. The reaction mixture was poured into saturated NaHCO₃ (300 mL) and extracted with CH₂Cl₂ (3 × 200 mL). The extracts were combined, washed with water (100 mL), dried, and evaporated. The residue was chromatographed (preparative HPLC, silica gel, 3:1 hexane/EtOAc) to afford 1.9 g (48%) of a 5:1 mixture of 24 and the β -anomer 32: IR 5.78, 5.90 (C=O), 6.95, 7.35, 8.28, 9.64, 13.95 μm; NMR δ 1.88 (s, 3, OAc), 2.08 (s, 3, OAc), 2.32 (s, 3, ArMe), 3.30 (s, 2.5, 2-OMe of α -anomer), 3.48 (s, 0.5, 2-OMe of β -anomer), 3.76 (s, 3, ArOMe), 4.0–4.6 (m, 4), 4.86 (t, 1, 3-H), 5.52 (m, 1, 4-H), 5.59 (d, 1, *J* = 3 Hz, 1-H), 6.67 (m, 2, Ar 3- and 5-H's), 7.10 (d, 1, *J* = 8 Hz, Ar 6-H), 7.78 (m, 4, Pht H's); TLC (2:1 hexane/EtOAc) *R_f* 0.30.

4-Methoxy-2-methylphenyl 3-Deoxy-2-methoxy-3-phthalimido- α -D-glucopyranoside (25). To a solution of crude 23 (300 mg) in MeOH (30 mL) was added Dowex 2X8 resin (OH⁻ form, 300 mg). The mixture was refluxed for 0.5 h, filtered, and evaporated. The residue was recrystallized from CH₂Cl₂/hexane and chromatographed (PLC, 97:3 CH₂Cl₂/MeOH) to afford 25 as a 5:1 α/β anomeric mixture in 32% yield from 18: IR 2.88 (OH), 5.63, 5.84 (Pht C=O), 6.22, 6.70, 7.30, 8.30, 9.05, 9.40, 9.75, 10.65, 10.95, 11.50, 12.60, 13.90 μm; NMR δ 2.26 (s, 3, ArMe), 3.28 (s, 2.5, 2-OMe of α -anomer), 3.46 (s, 0.5, 2-OMe of β -anomer), 3.76 (s, 3, ArOMe), 3.75–4.00 (m, 4), 4.30 (dd, 1, 2-H), 4.80 (t, 1, 3-H), 5.57 (d, 1, *J* = 3 Hz, 1-H), 6.6–6.8 (m, 2, Ar 3- and 5-H's), 7.06 (d, 1, Ar 6-H), 7.6–7.9 (m, 4, Pht H's); UV (MeOH) λ_{max} 285–289 nm (ε 41 27), 220.5 (51 100); MS, *m/e* (relative intensity) 443 M⁺ (2), 306 (38), 288 (1), 274 (5), 258 (63), 256 (18), 226 (28), 216 (26), 138 (100); TLC (97:3 CH₂Cl₂/MeOH) *R_f* 0.20.

Anal. Calcd for C₂₃H₂₅NO₈·0.2H₂O: C, 61.79; H, 5.73; N, 3.13. Found: C, 61.70; H, 5.87; N, 3.09.

Method B. To a solution of 24 (471 mg, 0.89 mmol) in MeOH (235 mL) was added saturated K₂CO₃ in MeOH (0.8 mL). The solution was stirred at 23 °C for 2 h, neutralized with HOAc in MeOH, and evaporated. The residue was triturated with CHCl₃ and filtered and the filtrate evaporated to afford 410 mg (100%) of 25.

4-Methoxy-2-methylphenyl 3-Deoxy-6-[(methylsulfonyloxy]-2-methoxy-3-phthalimido- α -D-glucopyranoside (26). To a solution of 25 (2.22 g, 5.0 mmol) in pyridine (20 mL) at 0 °C was added methanesulfonyl chloride (0.44 mL, 5.7 mmol). The mixture was stirred at 0 °C for 48 h, poured into water (100 mL), and extracted with CH₂Cl₂ (2 × 150 mL). The extracts were combined, washed with saturated NaHCO₃ (50 mL) and water (50 mL), dried, and evaporated to afford 2.54 g (98%) of 26: IR 2.86 (OH), 5.62, 5.89 (Pht C=O), 6.22, 6.70, 6.89, 7.18, 7.31, 7.45, 8.30, 8.57, 9.03, 9.18, 9.38, 9.58, 9.78, 10.10, 10.88, 11.50, 12.05, 13.50, 13.90 μm; NMR δ 2.35 (s, 3, ArMe), 3.07 (s, 3, SO₂Me), 3.29 (s, 3, 2-OMe), 3.77 (s, 3, ArOMe), 4.09 (dt, 1, 5-H), 4.35 (dd, 2), 4.45 (dd, 1), 4.60 (dd, 1), 4.80 (t, 1, 3-H), 5.60 (d, 1, *J* = 3 Hz, 1-H), 6.68 (dd, 1, Ar 5-H), 6.75 (d, 1, Ar 3-H), 7.10 (d, 1, Ar 6-H), 7.6–7.9 (m, 4, Pht H's); TLC (98:2 CH₂Cl₂/MeOH) *R_f* 0.30.

Anal. Calcd for C₂₄H₂₇NO₁₀S·0.3H₂O: C, 54.70; H, 5.28; N, 2.66. Found: C, 54.56; H, 5.36; N, 2.63.

4-Methoxy-2-methylphenyl 3,6-Dideoxy-6-iodo-2-methoxy-3-phthalimido- α -D-glucopyranoside (27). 26 (2.0 g, 3.84 mmol) was placed in 10% KI in acetone (80 mL) and stirred at 50 °C for 120 h. The mixture was filtered and the filtrate evaporated. The residue was partitioned between CH₂Cl₂ (50 mL) and H₂O (20 mL). The organic layer was separated, dried, and evaporated to afford 2.10 g (97%) of 27: IR 2.90 (OH), 5.60, 5.85

(C=O), 6.70, 6.88, 7.30, 8.30, 9.05, 9.40, 9.70, 10.60, 10.90, 11.50, 13.55 μm ; NMR δ 2.30 (s, 3, ArMe), 3.25 (s, 3, 2-OMe), 3.3–3.7 (m, 2), 3.75 (s, 3, ArOMe), 4.1–4.4 (m, 2), 4.80 (t, 1, 3-H), 5.55 (d, 1, $J = 3$ Hz, 1-H), 6.6–6.8 (m, 2, Ar 3- and 5-H's), 7.20 (m, 2, Ar 6-H), 7.6–7.8 (m, 4, Pht H's); TLC (99:1 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) R_f 0.30.

4-Methoxy-2-methylphenyl 4-O-Acetyl-3,6-dideoxy-6-iodo-2-methoxy-3-phthalimido- α -D-glucopyranoside (28). To a solution of 27 (2.0 g, 3.55 mmol) and 4-(dimethylamino)pyridine (80 mg, 0.65 mmol) in CH_2Cl_2 (20 mL) was added acetic anhydride (0.64 mL, 6.70 mmol). The solution was stirred at 23 °C for 2 h, diluted with CH_2Cl_2 (40 mL), washed with 1 M citric acid, saturated NaHCO_3 , and water, dried, and evaporated. The residue was chromatographed (dry column, silica gel, 98.5:1.5 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) to afford 1.6 g (80%) of 28: IR 5.70, 5.80 (C=O), 6.65, 6.85, 7.25, 8.20, 9.05, 9.55, 10.45, 11.40, 11.90, 12.50, 13.85 μm ; NMR δ 1.90 (s, 3, OAc), 2.30 (s, 3, ArMe), 3.25 (s, 3, 2-OMe), 3.70 (s, 3, ArOMe), 3.8–4.1 (m, 1), 4.50 (dd, 1, 2-H), 4.90 (t, 1, 3-H), 5.40 (t, 1, 4-H), 5.60 (d, 1, $J = 3$ Hz, 1-H), 6.6–6.8 (m, 2, Ar 3- and 5-H's), 7.30 (d, 1, Ar 6-H), 7.6–7.9 (m, 4, Pht H's); TLC (99:1 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) R_f 0.7.

Anal. Calcd for $\text{C}_{25}\text{H}_{28}\text{INO}_8$: C, 50.43; H, 4.40; N, 2.35. Found: C, 50.94; H, 4.79; N, 2.27.

4-Methoxy-2-methylphenyl 3-Deoxy-6-[(methylsulfonyl)oxy]-2,4-dimethoxy-3-phthalimido- α -D-glucopyranoside (29). A mixture of 26 (229.9 mg, 0.44 mmol), Ag_2O (600 mg), and methyl iodide (2.0 mL) in CH_2Cl_2 (12.0 mL) was refluxed for 4 h. The reaction mixture was filtered through Celite and evaporated. The residue was crystallized from $\text{CHCl}_3/\text{hexane}$ to afford 201.4 mg (85%) of 29: IR 5.62, 5.83 (Pht C=O), 6.69, 6.86, 7.28, 7.39, 8.27, 8.50, 9.10, 9.69, 13.85 μm ; NMR δ 2.35 (s, 3, ArMe), 3.09 (s, 3, SO_2Me), 3.30 and 3.35 (2 s, 6, 2- and 4-OMe's), 3.78 (s, 3, ArOMe), 4.0–4.5 (m, 5), 4.88 (t, 1, 3-H), 5.57 (d, 1, $J = 3$ Hz, 1-H), 6.67 (dd, 1, Ar 5-H), 6.71 (br s, 1, Ar 3-H), 7.08 (d, 1, Ar 6-H), 7.80 (m, 4, Pht H's); TLC (95:5 $\text{CHCl}_3/\text{MeOH}$) R_f 0.73.

Anal. Calcd for $\text{C}_{25}\text{H}_{29}\text{NO}_{10}\text{S}$: C, 56.07; H, 5.45; N, 2.61. Found: C, 55.81; H, 5.27; N, 2.47.

4-Methoxy-2-methylphenyl 3,6-Dideoxy-6-iodo-2,4-dimethoxy-3-phthalimido- α -D-glucopyranoside (30). A solution of 29 (52.4 mg, 0.098 mmol) and NaI (300 mg) in 2-butanone (3.0 mL) was refluxed for 6 h, filtered, and evaporated. The residue was partitioned between CHCl_3 (15 mL) and water (5 mL). The organic solution was separated, washed with water (5 mL) and saturated NaCl (5 mL), dried, and evaporated. The residue was taken up in 1:1 hexane/EtOAc (5 mL) and passed through a silica gel column (1 \times 3 cm). The eluent was evaporated to afford 55.5 mg (100%) of 30 as a syrup: IR 5.67, 5.88 (Pht C=O), 6.25, 6.73, 6.92, 7.35, 7.86, 8.30, 8.55, 9.08, 9.25, 9.75, 10.10, 10.60, 11.50, 12.50, 13.90 μm ; NMR δ 2.32 (s, 3, ArMe), 3.30 and 3.40 (2 s, 6, 2- and 4-OMe's), 3.45–3.70 (m, 3), 3.79 (s, 3, ArOMe), 4.01 (t, 1, 4-H), 4.29 (dd, 1, $J = 2$ Hz and 10 Hz, 2-H), 4.90 (t, 1, 3-H), 5.58 (d, 1, $J = 2$ Hz, 1-H), 6.68 (m, 2, Ar 3- and 5-H's), 7.08 (d, 1, Ar 6-H), 7.82 (m, 4, Pht H's); TLC (1:1 hexane/EtOAc) R_f 0.63.

3,5-Dimethoxyphenyl 4,6-Di-O-acetyl-3-deoxy-2-methoxy-3-phthalimido- β -D-glucopyranoside (33). SnCl_4 (1.7 mL) was added dropwise over 0.5 h to a solution of 21 (5.0 g, 11.14 mmol) and 3,5-dimethoxyphenol (3.4 g, 22.28 mmol) in CH_2Cl_2 (50 mL) at 0 °C. The mixture was stirred at 23 °C for 8 h. The reaction mixture was poured into saturated NaHCO_3 (200 mL) and extracted with CH_2Cl_2 (3 \times 200 mL). The extracts were combined, washed with water (150 mL), dried, and evaporated. The residue was chromatographed (preparative HPLC, silica gel, 3:1 hexane/EtOAc) to afford 1.95 g (31%) of 33: IR 5.78, 5.88 (C=O), 6.26, 6.40, 6.90, 7.32, 7.68, 8.00, 8.18, 8.32, 8.70, 9.00, 9.15, 9.60, 9.68, 11.45, 12.35, 13.90 μm ; NMR δ 1.88 (s, 3, OAc), 2.08 (s, 3, OAc), 2.81 (s, 3, 2-OMe), 3.78 (s, 6, ArOMe's), 3.91 (m, 1), 4.27 (m, 2), 4.32 (m, 2), 5.02 (d, 1, $J = 8$ Hz, 1-H), 5.68 (t, 1, 4-H), 6.08 (m, 2, Ar H's), 7.52 (br s, 1, Ar H), 7.6–7.9 (m, 4, Pht H's); TLC (2:1 hexane/EtOAc) R_f 0.25.

Anal. Calcd for $\text{C}_{27}\text{H}_{29}\text{NO}_{11}$: C, 59.66; H, 5.37; N, 2.58. Found: C, 59.44; H, 5.08; N, 2.43.

4-Methoxy-2-methylphenyl 4-O-Acetyl-3-deoxy-2-methoxy-3-phthalimido- α -D-xylo-hex-5-enopyranoside (35). To a solution of 28 (1.55 g, 2.61 mmol) in HMPA (8.0 mL) was added DBU (0.62 mL, 4.0 mmol). The solution was stirred at 50 °C for 60 h, diluted with ether (150 mL), and filtered. The filtrate was washed with 10% NaHSO_4 (50 mL), saturated NaHCO_3 (50 mL),

and water (50 mL), dried, and evaporated. The residue was chromatographed (PLC, silica gel, 99.5:0.5 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) to afford 860 mg (71%) of 35: IR 5.75, 5.90 (Pht C=O), 6.70, 6.90, 7.20, 7.35, 8.30, 8.80, 9.05, 9.45, 9.65, 9.85, 10.50, 11.50, 12.05, 12.35, 13.50, 13.95 μm ; NMR δ 1.90 (s, 3, OAc), 2.30 (s, 3, ArMe), 3.30 (s, 3, 2-OMe), 3.70 (s, 3, ArOMe), 4.5–4.7 (m, 3), 4.90 (t, 1, 3-H), 5.60 (d, 1, $J = 3$ Hz, 1-H), 6.10 (dt, 1, 4-H), 6.5–6.8 (m, 2, Ar 3- and 5-H's), 7.6–7.9 (m, 4, Pht H's); TLC (99:1 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) R_f 0.7.

Anal. Calcd for $\text{C}_{25}\text{H}_{25}\text{NO}_8$: C, 64.23; H, 5.39; N, 2.99. Found: C, 63.76; H, 5.78; N, 3.08.

4-Methoxy-2-methylphenyl 3-Deoxy-2,4-dimethoxy-3-phthalimido- α -D-xylo-hex-5-enopyranoside (36). By a procedure similar to that described above 30 (79.8 mg, 0.14 mmol) was treated with DBU (42.7 mg, 0.28 mmol) in HMPA (1.0 mL). The crude product was chromatographed (PLC, silica gel, 2:1 hexane/EtOAc) to afford 35.5 mg (58%) of 36: IR 5.61, 5.82, 5.95, 6.70, 6.90, 7.30, 7.81, 8.32, 8.74, 9.00, 9.10, 9.32, 9.55, 9.85, 10.10, 10.55, 11.58, 12.00, 12.55, 13.95 μm ; NMR δ 2.30 (s, 3, ArMe), 3.32 and 3.41 (2 s, 6, 2- and 4-OMe's), 3.79 (s, 3, ArOMe), 4.3–4.9 (m, 5), 5.64 (d, 1, $J = 3$ Hz, 1-H), 6.69 (m, 2, Ar 3- and 5-H's), 7.17 (d, 1, Ar 6-H), 7.82 (m, 4, Pht H's); TLC (99.4:0.6 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) R_f 0.6.

Anal. Calcd for $\text{C}_{24}\text{H}_{25}\text{NO}_7$: C, 65.59; H, 5.73; N, 3.19. Found: C, 65.15; H, 5.75; N, 3.20.

Cation-Induced Cyclization Studies. a. Reaction of 35 with Iodonium Dicolloidine Perchlorate in the Presence of EtOH. Freshly prepared iodonium dicollidine perchlorate (46.7 mg, 0.1 mmol) was added to a solution of 35 (46.7 mg, 0.1 mmol) in 9:1 $\text{CHCl}_3/\text{EtOH}$ (0.7 mL), and the mixture was stirred in the dark for 16 h. Additional iodonium dicollidine perchlorate (0.1 mmol) was added, and stirring was continued for 6 h. The excess reagent was precipitated by addition of ether (20 mL), and the filtrate was washed with 0.5 M $\text{Na}_2\text{S}_2\text{O}_4$ (5 mL), saturated NaHCO_3 (5 mL), and water (5 mL), dried, and evaporated. The residue was chromatographed (PLC, silica gel, CHCl_3) to afford 10 mg (16%) of 40 as the major product: NMR δ 1.22 (t, 3, OCH_2CH_3), 1.92 (s, 3, ArMe), 2.30 (s, 3, 2-OMe), 3.22 (s, 3, OAc), 3.5–3.9 (m, 7, ArOMe, OCH_2CH_3 , 6- H_2), 4.63 (dd, 1, $J = 8$ Hz and $J = 2$ Hz, 2-H), 4.92 (t, 1, $J = 8$ Hz, 3-H), 5.68 (d, 1, $J = 2$ Hz, 1-H), 5.97 (d, 1, $J = 8$ Hz, 4-H), 6.70 (m, 2, Ar H's), 7.32 (m, 1, Ar H), 7.6–7.9 (m, 4, Pht H's).

b. Reaction of 36 with Trifluoroacetic Acid. 36 (26 mg, 0.055 mmol) and trifluoroacetic acid (1.6 μL , 0.021 mmol) were placed in CDCl_3 (0.4 mL) and allowed to stand at 23 °C for 16 h at which time no 36 could be detected by NMR. The reaction mixture was diluted with CHCl_3 (10 mL), washed with saturated NaHCO_3 (5 mL), dried, and evaporated to afford a complex (TLC) mixture. The residue was chromatographed (PLC, silica gel, 99.7:0.3 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) to afford 4 mg of 36, identical with an authentic sample by MS, NMR, and TLC, as the major product. Several experiments gave similar results with 36 always the principal product accompanied by low (0–5%) and erratic yields of aglycon 9 and materials tentatively identified (MS) as 38 and 42.

c. Reaction of 36 with N-Bromosuccinimide. *N*-Bromosuccinimide (9.7 mg, 0.055 mmol) and 36 (26.0 mg, 0.055 mmol) were placed in CDCl_3 (0.5 mL) and allowed to stand at 23 °C for 5 min. The reaction mixture was diluted with CHCl_3 (10 mL), washed with water (5 mL), dried, and evaporated. The residue was chromatographed (PLC, silica gel, 99.75:0.25 $\text{CH}_2\text{Cl}_2/\text{MeOH}$) to afford 5 mg of 41 as the major component: MS, m/e 616 (M^+); NMR δ 2.20 (s, 3, ArMe), 2.88 (s, 4, succinimide H's), 3.22, 3.31 (2 s, 6, 2- and 4-OMe's), 3.68 (s, 3, ArOMe), 4.3–4.9 (m, 5), 5.58 (d, 1, $J = 2$ Hz, 1-H), 6.60 (m, 2, Ar H's), 7.03 (m, 1, Ar H), 7.6–7.9 (m, 4, Pht H's).

d. Reaction of 36 with Iodonium Dicolloidine Perchlorate. Freshly prepared iodonium dicollidine perchlorate (16.8 mg, 0.036 mmol) and 36 (15.6 mg, 0.036 mmol) were placed in CDCl_3 (0.5 mL) and allowed to stand at 23 °C for 0.5 h. Ether (20 mL) was added to precipitate excess reagent. The filtrate was washed with 0.5 M $\text{Na}_2\text{S}_2\text{O}_4$ (5 mL), saturated NaHCO_3 (5 mL), and water (5 mL), dried, and evaporated. The residue was chromatographed (PLC, silica gel, CHCl_3 , three passes) to afford 7.0 mg (34%) of 43: NMR δ 2.23 (s, 3, ArMe), 3.28, 3.36 (2 s, 6, 2- and 4-OMe's), 3.72 (s, 3, ArOMe), 4.38 (dd, 1, $J = 8$ Hz and $J = 2$ Hz, 2-H), 4.58

(dd, 1, $J = 8$ Hz and $J = 1$ Hz, 4-H), 4.90 (t, 1, $J = 8$ Hz, 3-H), 5.70 (m, 2, 1- and 6-H's), 6.60 (m, 2, Ar H's), 7.25 (m, 1, Ar H), 7.6-7.9 (m, 4, Pht H's); MS, m/e 565 (M^+); TLC ($CHCl_3$, two passes) R_f 0.42.

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Synthesis of Bufalitoxin and Bufotoxin^{1a,2}

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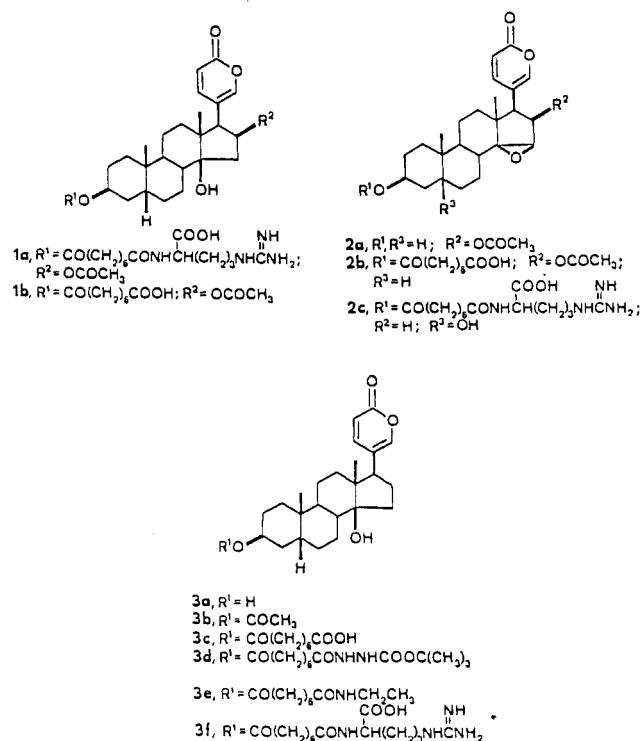
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Completion of a formal total synthetic route to the toad venom constituents bufotalin (8), cinobufagin (2a), bufalitoxin (3f), and bufotoxin (1a) has been accomplished. Bufalin (3a) was employed as relay and converted to 14-dehydrobufalin 3-acetate (4). Selective oxidation of olefin 4 with a chromium trioxide-pyridine reagent afforded 16-ketone 5, which we had previously transformed to bufotalin (8) and thence to cinobufagin (2a). Condensation of bufalin (3a) with suberic anhydride followed by a mixed carbonic anhydride reaction sequence using arginine monohydrochloride yielded bufalitoxin (3f), and an analogous route from bufotalin (8) led to bufotoxin (1a).

In a classic investigation of toad venom constituents, Wieland and colleagues isolated bufotoxin (1a) in 1922³ from the European toad *Bufo vulgaris* (*Bufo bufo bufo* Linné), and some 20 years later they were able to propose a tentative structure.⁴ Meanwhile, Kondo and co-workers succeeded in isolating bufotoxin and the parent steroid bufotalin from the Japanese toad venom preparation Senso (the Chinese Ch'an Su).⁵ The same substance was reisolated from *Bufo bufo bufo* L. and named vulgarobufotoxin.⁶ In 1955 the isolation of bufotoxin from *Bufo bufo* L. was reconfirmed by the Reichstein group⁷ and more recently the correct structure proposed by one of us (Y.K.) and Meyer⁸ was confirmed by our partial synthesis of bufotoxin from bufotalin.²

Registry No. 9, 5307-05-1; 12, 4603-89-8; 13, 42775-84-8; 14, 108512-22-7; 15, 108590-49-4; 16, 108590-50-7; 17, 108512-23-8; 18, 108512-24-9; 20, 108512-25-0; 21, 108512-26-1; 23, 108512-27-2; 24, 108512-29-4; 25, 108512-31-8; 26, 108512-32-9; 27, 108512-33-0; 28, 108512-34-1; 29, 108512-35-2; 30, 108512-36-3; 31, 108512-28-3; 32, 108512-30-7; 33, 108512-37-4; 35, 108512-38-5; 36, 108512-39-6; 38, 108512-41-0; 40, 108512-40-9; 41, 108512-43-2; 42, 108512-42-1; 43, 108512-44-3; *N*-(ethoxycarbonyl)phthalimide, 22509-74-6; 3,5-dimethoxyphenol, 500-99-2.



(1) (a) The present contribution is part 104 of Steroids and Related Natural Products and series number 36 of Bufadienolides; for parts 103 and 35 refer to, respectively: Pettit, G. R.; Herald, D. L.; Herald, C. L.; Kokke, W. C. M. C.; Djerassi, C. *Steroids* 1986, 47, 321. Green, B.; Sntzke, F.; Sntzke, G.; Pettit, G. R.; Kamano, Y.; Niven, M. L. *Croat. Chim. Acta* 1985, 58, 371. (b) On leave from the Institute of Organic Chemistry and Biochemistry, Czechoslovak Academy of Sciences, 166 10 Prague, Czechoslovakia.

(2) A portion of the present study was summarized in a preliminary report by: Pettit, G. R.; Kamano, Y. *Chem. Commun.* 1972, 45.

(3) Wieland, H.; Alles, R. *Chem. Ber.* 1922, 55, 1789.

(4) Wieland, H.; Behringer, H. *Ann.* 1941, 549, 209. In this remarkable study the average yield of crude bufotoxin from *Bufo vulgaris* (800 males and 400 females) was found to be 1.34 mg/toad, while the female was found to yield 1.23 mg on the average of bufotalin. The male provided only 0.55 mg of bufotalin. In the same investigation, arenobufotoxin was isolated from the South American toad *Bufo arenarum* and purified by column chromatography on aluminum oxide.

(5) Kondo, H.; Ikawa, S. *J. Pharm. Soc., Jpn.* 1933, 53, 23; *Chem. Abstr.* 1933, 27, 1887. Kondo, H.; Ono, S. *J. Pharm. Soc., Jpn.* 1938, 58, 37; *Chem. Abstr.* 1939, 32, 3765.

(6) Chen, K. K.; Jensen, H.; Chen, A. L. *J. Pharmacol.* 1933, 47, 307.

(7) Urscheler, H. R.; Tamm, C.; Reichstein, T. *Helv. Chim. Acta* 1955, 38, 883.

(8) For leading citations refer to footnote 5 of ref 2. See also: Meyer, K.; Linde, H. In *Venomous Animals and Their Venoms*; Bücherl, W., Buckley, E. E., Eds.; 1971; Vol. 2, p 521. For more recent advances in the chemistry of naturally occurring bufadienolides, refer to: Ode, H.; Kamano, Y.; Pettit, G. R. *MTP International Review of Science, Organic Chemistry Series One*; Johns, W. F., Ed.; Butterworths: London, 1972; Vol. 8, Chapter 6, pp 151-177. Ode, R. H.; Pettit, G. R.; Kamano, Y. *MTP International Review of Science, Organic Chemistry Series Two*; Johns, W. F., Ed.; Butterworths: London, 1975; Vol. 8, Chapter 6, pp 145-171. Nassimbeni, L. R.; Niven, M. L.; Sheldrick, G. M.; Pettit, G. R.; Inoue, M.; Kamano, Y. *Acta Crystallogr., Sect. C: Cryst. Struct. Commun.* 1983, C39, 801 and ref 1a.

Until 1974 the only known bufotoxin-type toad venom constituents were suberylarginine esters of a β -hydroxybufadienolide. Investigation of the venom from *Bufo vulgaris formosus* Boulenger has led to the isolation of bufotoxins with succinic, glutaric, pimelic, or adipic acid replacing suberic acid.^{9a} From *Bufo melanostictus*

(9) (a) Shimada, K.; Fujii, Y.; Mitsuishi, E.; Nambara, T. *Tetrahedron Lett.* 1974, 467. Shimada, K.; Fujii, Y.; Niizaki, Y.; Nambara, T. *Tetrahedron Lett.* 1975, 653. Shimada, K.; Sato, Y.; Fujii, Y.; Nambara, T. *Chem. Pharm. Bull.* 1976, 24, 1120. Shimada, K.; Ro, J. S.; Ohishi, K.; Nambara, T. *Chem. Pharm. Bull.* 1985, 33, 2767. (b) Shimada, K.; Fujii, Y.; Yamashita, E.; Niizaki, Y.; Sato, Y.; Nambara, T. *Chem. Pharm. Bull.* 1977, 25, 714. See also: Shimada, K.; Ohishi, K.; Nambara, T. *Tetrahedron Lett.* 1984, 24, 551. Shimada, K.; Ohishi, K.; Nambara, T. *Chem. Pharm. Bull.* 1984, 32, 4396. (c) Shimada, K.; Nambara, T. *Tetrahedron Lett.* 1979, 163. Shimada, K.; Nambara, T. *Chem. Pharm. Bull.* 1980, 28, 1559.